L_3 for a hydrogen atom without spin:

$$H|n, \ell, m\rangle = E_n |n, \ell, m\rangle$$

$$L^2|n, \ell, m\rangle = \hbar^2 \ell (\ell + 1) |n, \ell, m\rangle$$

$$L_3|n, \ell, m\rangle = \hbar m |n, \ell, m\rangle.$$
(1.415)

Suppose the states $|\sigma\rangle$ for $\sigma = \pm$ are the eigenstates of the third component S_3 of the operator **S** that represents the spin of the electron

$$S_3|\sigma\rangle = \sigma \frac{\hbar}{2}|\sigma\rangle. \tag{1.416}$$

Then the direct- or tensor-product states

$$|n,\ell,m,\sigma\rangle \equiv |n,\ell,m\rangle \otimes |\sigma\rangle \equiv |n,\ell,m\rangle |\sigma\rangle$$
(1.417)

represent a hydrogen atom including the spin of its electron. They are eigenvectors of all four operators H, L^2 , L_3 , and S_3 :

$$H|n,\ell,m,\sigma\rangle = E_n|n,\ell,m,\sigma\rangle \qquad L^2|n,\ell,m,\sigma\rangle = \hbar^2\ell(\ell+1)|n,\ell,m,\sigma\rangle$$

$$L_3|n,\ell,m,\sigma\rangle = \hbar m|n,\ell,m,\sigma\rangle \qquad S_3|n,\ell,m,\sigma\rangle = \sigma \frac{\hbar}{2}|n,\ell,m,\sigma\rangle.$$
(1.418)

Suitable linear combinations of these states are eigenvectors of the square J^2 of the composite angular momentum J = L + S as well as of J_3 , L_3 , and S_3 .

Example 1.51 (Adding Two Spins) The smallest positive value of angular momentum is $\hbar/2$. The spin-one-half angular momentum operators S are represented by three 2×2 matrices

$$S_a = \frac{\hbar}{2} \,\sigma_a \tag{1.419}$$

in which the σ_a are the Pauli matrices

$$\sigma_1 = \begin{pmatrix} 0 & 1 \\ 1 & 0 \end{pmatrix}, \quad \sigma_2 = \begin{pmatrix} 0 & -i \\ i & 0 \end{pmatrix}, \quad \text{and} \quad \sigma_3 = \begin{pmatrix} 1 & 0 \\ 0 & -1 \end{pmatrix}. \tag{1.420}$$

Consider two spin operators $S^{(1)}$ and $S^{(2)}$ acting on two spin-one-half systems. The states $|\pm\rangle_1$ are eigenstates of $S_3^{(1)}$, and the states $|\pm\rangle_2$ are eigenstates of $S_3^{(2)}$

$$S_3^{(1)}|\pm\rangle_1 = \pm \frac{\hbar}{2}|\pm\rangle_1$$
 and $S_3^{(2)}|\pm\rangle_2 = \pm \frac{\hbar}{2}|\pm\rangle_2.$ (1.421)

72