

1. Compute the matrix element of the third component z of the position operator r between the $3p$ state with $m = 0$ and the ground state of atomic hydrogen:

$$\langle 100|z|310\rangle. \quad (1)$$

We use

$$\langle r, \theta, \phi | 100 \rangle = \frac{1}{\sqrt{\pi}} \frac{1}{a_0^{3/2}} e^{-r/a_0}$$

and

$$\langle r, \theta, \phi | 310 \rangle = \frac{1}{81} \sqrt{\frac{2}{\pi}} \frac{1}{a_0^{3/2}} \left(6 - \frac{r}{a_0}\right) \frac{r}{a_0} e^{-r/3a_0} \cos\theta$$

where

$$a_0 = \frac{\hbar^2}{m_e e^2} = \frac{1}{\alpha} \frac{\hbar c}{m_e c^2} \cong 0.53 \text{ \AA}.$$

Now $z = r \cos\theta$, so

$$\langle 100|z|310\rangle = \int_0^\infty r^3 dr \int_{-1}^1 dx \int_{2\pi} d\phi (r x^2) \frac{\sqrt{2}}{81 \pi a_0^3} \left(6 - \frac{r}{a_0}\right) \frac{r}{a_0} x e^{-r/3a_0} e^{-r/a_0}$$

$$= \frac{2\sqrt{2}}{81} a_0^3 \int_0^\infty dr r^3 \left(6 - \frac{r}{a_0}\right) \frac{r}{a_0} e^{-\frac{4r}{3a_0}} \int_{-1}^1 x^2 dx$$

$$= \frac{4\sqrt{2}}{3 \cdot 81} \frac{1}{a_0^3} \int_0^\infty \frac{r^4}{a_0} \left(6 - \frac{r}{a_0}\right) e^{-4r/3a_0} dr.$$

Let $u = v/a_0$. Then

$$\langle 100 | z | 310 \rangle = \frac{2^2 \sqrt{2} a_0}{3^5} \int_0^\infty u^4 (6-u) e^{-4u/3} du$$

$$= \frac{2^2 \sqrt{2} a_0}{3^5} \left[6 \int_0^\infty u^4 e^{-4u/3} du - \int_0^\infty u^5 e^{-4u/3} du \right]$$

$$= \frac{2^2 \sqrt{2}}{3^5} \left(6 \frac{4! 3^5}{4^5} - \frac{5! 3^6}{4^6} \right) a_0$$

$$= \frac{2^2 \sqrt{2}}{3^5} \left(\frac{6 \cdot 3 \cdot 2}{2^8} - \frac{5 \cdot 3 \cdot 2 \cdot 3}{2^{10}} \right) a_0$$

$$= \frac{2^2 \sqrt{2}}{3^5} \left(\frac{3^2}{2^6} - \frac{5 \cdot 3^2}{2^9} \right) a_0$$

$$= \frac{\sqrt{2} 3^2}{2^4} \left(1 - \frac{5}{2^3} \right) a_0 = \frac{\sqrt{2} 3^2}{2^4} \left(\frac{2^3 - 5}{2^3} \right) a_0$$

$$= \frac{\sqrt{2} 3^2}{2^4} \left(\frac{3}{8} \right) a_0 = \frac{\sqrt{2} 3^3}{2^7} a_0$$

$$= \frac{27}{64\sqrt{2}} a_0 = \frac{27}{64\sqrt{2}} \frac{1}{\alpha} \frac{\hbar c}{m_e c^2}$$

2. Compute the rate of transitions from the $3p$ state with $m = 0$ (directly) to the ground state of atomic hydrogen. What is the partial lifetime of the $3p$ state, excluding decays to the $2s$ state?

The amplitude is

$$\langle 100 | S(\epsilon, 0) | 310 \rangle = \frac{q}{m} \left(\frac{\hbar}{2\epsilon_0 V \omega_k} \right)^{\frac{1}{2}} \langle 100 | \epsilon^* \cdot \vec{p} | 310 \rangle e^{-i\Delta E \epsilon / \hbar}$$

in the dipole approximation. So the probability is

$$P(\epsilon) = |\langle 100 | S(\epsilon, 0) | 310 \rangle|^2 = \left(\frac{q}{m} \right)^2 \frac{\hbar}{2\epsilon_0 V \omega_k} |\langle 100 | \epsilon^* \cdot \vec{p} | 310 \rangle|^2 \frac{2\pi\epsilon}{\hbar} \delta(\Delta E)$$

and the rate is

$$\hat{w} = \frac{2\pi}{\hbar} \left(\frac{q}{m} \right)^2 \frac{\hbar}{2\epsilon_0 V \omega_k} |\langle 100 | \epsilon^* \cdot \vec{p} | 310 \rangle|^2 \delta(\Delta E)$$

where $\Delta E = E_{100} + \hbar\omega_k - E_{310}$. Now

$$\vec{p} = \frac{m}{i\hbar} [x, H_0]$$

so the matrix element is

$$\begin{aligned} |\langle 100 | \epsilon^* \cdot \vec{p} | 310 \rangle|^2 &= \frac{m^2}{\hbar^2} (E_{310} - E_{100})^2 |\langle 100 | \epsilon^* \cdot x | 310 \rangle|^2 \\ &= \frac{m^2}{\hbar^2} (E_{310} - E_{100})^2 |\langle 100 | z | 310 \rangle|^2 |\epsilon_{r3}|^2 \end{aligned}$$

From problem (1), we have

$$\langle 100 | \hat{z} | 310 \rangle = \frac{27}{64\sqrt{2}} \frac{1}{\alpha} \frac{\hbar c}{mc^2}$$

And

$$(E_{310} - E_{100})^2 = \left(\frac{1}{2} mc^2 \alpha^2 \right)^2 \left(1 - \frac{1}{3^2} \right)^2 = \frac{16}{81} (mc^2 \alpha^2)^2$$

And $\delta(\Delta E)$ ensures that

$$\hbar \omega_k = E_{310} - E_{100} = \frac{4}{9} mc^2 \alpha^2$$

So since $q^2 = 4\pi\epsilon_0 \hbar c \alpha$, the rate \hat{W} is

$$\begin{aligned} \hat{W} &= \frac{2\pi}{\hbar} \frac{4\pi\epsilon_0 \hbar c \alpha}{m^2} \frac{\hbar^9 \hbar}{2\epsilon_0 V 4mc^2 \alpha^2} \frac{m^2}{\hbar^2} \frac{16 (mc^2 \alpha^2)^2}{81} \delta(\Delta E) \\ &\quad \times \left(\frac{27}{64\sqrt{2}} \frac{1}{\alpha} \frac{\hbar c}{mc^2} \right)^2 \epsilon_{r3}^*(k) \cdot \epsilon_{r3}(k) \\ &= \frac{3^4}{2^9} \frac{\pi^2 \hbar^2 c^2 \alpha}{mV} \epsilon_{r3}^*(k) \cdot \epsilon_{r3}(k) \delta(\Delta E). \end{aligned}$$

Now

$$\begin{aligned} \sum_{r=1}^2 \epsilon_{r3}^*(k) \cdot \epsilon_{r3}(k) &= \hat{z}^T \left(1 - \hat{h} \hat{h}^T \right) \hat{z} \\ &= 1 - (\hat{z} \cdot \hat{h})^2 = 1 - \cos^2 \theta = \sin^2 \theta. \end{aligned}$$

We now integrate over final states

$$W = V \int \frac{d^3k}{(2\pi)^3} \hat{W}$$

$$= V \int_0^\infty \frac{d\hbar\omega}{(2\pi)^2} \frac{k^2}{\hbar c} \int_{-1}^1 dx \frac{3^4 \pi^2 \hbar^2 c^2 \alpha}{2^9 m V} (1-x^2) \delta(\Delta E)$$

$$\int_{-1}^1 dx (1-x^2) = 2 - \left[\frac{x^3}{3} \right]_{-1}^1 = 2 - \left(\frac{1}{3} + \frac{1}{3} \right) = \frac{4}{3}$$

So

$$W = \frac{1}{(2\pi)^2} \frac{\hbar^2}{\hbar c} \frac{3^4 \pi^2 \hbar^2 c^2 \alpha}{2^9 m} \frac{4}{3}$$

$$= \frac{3^3}{2^9} \frac{\hbar^2 k^2 \alpha}{m} = \frac{3^3}{2^9} \frac{\alpha (\hbar\omega)^2}{m \hbar c^2}$$

$$= \frac{3^3}{2^9} \frac{\alpha}{m \hbar c^2} \left(\frac{4 m c^2 \alpha^2}{9} \right)^2$$

$$= \frac{1}{3 \cdot 2^5} \frac{m c^2 \alpha^5}{\hbar} \quad \text{and the partial}$$

lifetime is

$$\tau = 3 \cdot 2^5 \frac{\hbar}{m c^2} \alpha^{-5} = 3 \cdot 2^5 \frac{6.582 \times 10^{-22} (137.036)^5}{0.511} \text{ s}$$

That is,

$$\tau \approx 5.98 \text{ ns}$$

is an upper bound on the lifetime of the $3p$ state of atomic hydrogen.

The actual lifetime of the $3p$ state is

$$\tau = 5.41 \pm 0.18 \text{ ns}$$

as reported by Eberhart et al., Phys. Rev. A2, 2177-2179 (1970).